## ADDITIONS AND CORRECTIONS

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Alexander N. Tarnovsky,\* Villy Sundström, Eva Åkesson,\* and Torbjörn Pascher: Photochemistry of Diiodomethane in Solution Studied by Femtosecond and Nanosecond Laser Photolysis. Formation and Dark Reactions of the  $CH_2I-I$  Isomer Photoproduct and its Role in Cyclopropanation of Olefins

There were typographical errors in this paper regarding second-order rate constants for the cyclopropanation reaction between the CH<sub>2</sub>I–I isomer and cyclohexene in the solvents dichloromethane and acetonitrile. The correct second-order rate constants are (3.4  $\pm$  0.2)  $\times$  10<sup>7</sup> M $^{-1}$  s $^{-1}$  in CH<sub>2</sub>Cl<sub>2</sub> and (4.2  $\pm$  0.9)  $\times$  10<sup>7</sup> M $^{-1}$  s $^{-1}$  in CH<sub>3</sub>CN (instead of (3.4  $\pm$  0.2)  $\times$  10<sup>6</sup> and (4.2  $\pm$  0.9)  $\times$  10<sup>6</sup> M $^{-1}$  s $^{-1}$ , respectively).

This change affects the following places:

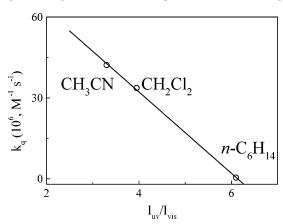
Abstract, page 237, last sentence.

Page 246, left column. "...(0.4–40)  $\times$  10<sup>6</sup> M<sup>-1</sup> s<sup>-1</sup>..." instead of "...(4–40)  $\times$  10<sup>5</sup> M<sup>-1</sup> s<sup>-1</sup>...".

Page 246. Caption of Figure 9, last sentence. The figure itself is correct.

Page 246. Table 3.

Page 246. Figure 10 should be replaced with the new figure.



**Figure 10.** The second-order rate constants for the cyclopropanation reaction between the  $CH_2I-I$  isomer and cyclohexene in the solvents  $CH_3CN$ ,  $n\text{-}C_6H_{14}$ , and  $CH_2Cl_2$  plotted vs the ratio between the amplitudes of the UV band ( $\sim$ 390 nm) and the visible CT band ( $\sim$ 560–570 nm) of  $CH_2I-I$  measured in the same solvents.

The conclusions in section 3.7 of the original manuscript are not altered: (i) it is the vibrationally relaxed CH<sub>2</sub>I–I isomer that reacts with cyclohexene; (ii) the second-order rate constants of this reaction increase significantly upon going from the solvent n-C<sub>6</sub>H<sub>14</sub> to CH<sub>2</sub>Cl<sub>2</sub> and CH<sub>3</sub>CN, but all are smaller than the diffusion-limited rates; (iii) the relative strength of the visible "iodide-to-carbenium" CT transition with respect to the  $\sim$ 390 nm transition of CH<sub>2</sub>I–I correlates with the second-order rate constants for the reaction between CH<sub>2</sub>I–I and cyclohexene in the solvents n-C<sub>6</sub>H<sub>14</sub>, CH<sub>2</sub>Cl<sub>2</sub>, and CH<sub>3</sub>CN (see Figure 10).

None of the corrections change any of the values in other tables, text, or figures.

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